

Advanced phosphorous and particle removal

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Importance of phosphorus

12. frequent element of the crust of earth

- Phosphorus is a chemical element. Symbol **P**
ordinal number 15 and has a specific weight of 31 g/mol
- Greek φωσ-φορος „light carrying“, given by the shining of the white phosphorus in the reaction with oxygen.
- P was discovered in 1669 by Hennig Brand, a German pharmacist and alchemist.
- Phosphorus is essential important for all organism.
 - ➡ P-compounds are parts of DNA- and RNA-molecules
 - ➡ The phosphorus containing compounds ADP/ATP are playing an important part in the energy metabolism of all cells.
- A man with 70 kg contains about 700 g phosphorus, from which 600 g is bound in the bones
- Important nutrient for the growth of plants and essential fertilizer
- Main resources: Morocco, South Africa, Russia, China and USA
Resource remain **only about 300 years**



Eutrophication

Effects of nutrient discharge

Nutrient (mainly N, P) emission



Increased algae growth („Algae bloom“)



Dead algae settle to the ground



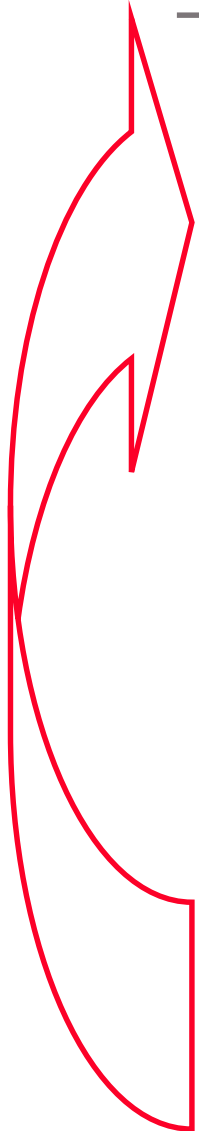
Degradation of algae with oxygen consumption



Oxygen depression/no O₂ available anymore



Mineralization and setting free of nutrients



Example of eutrophication in the Baltic Sea



Cyanobacterial blooms in the Baltic Sea
MODIS AQUA 2005-07-11, data from NASA
processed by SMHI

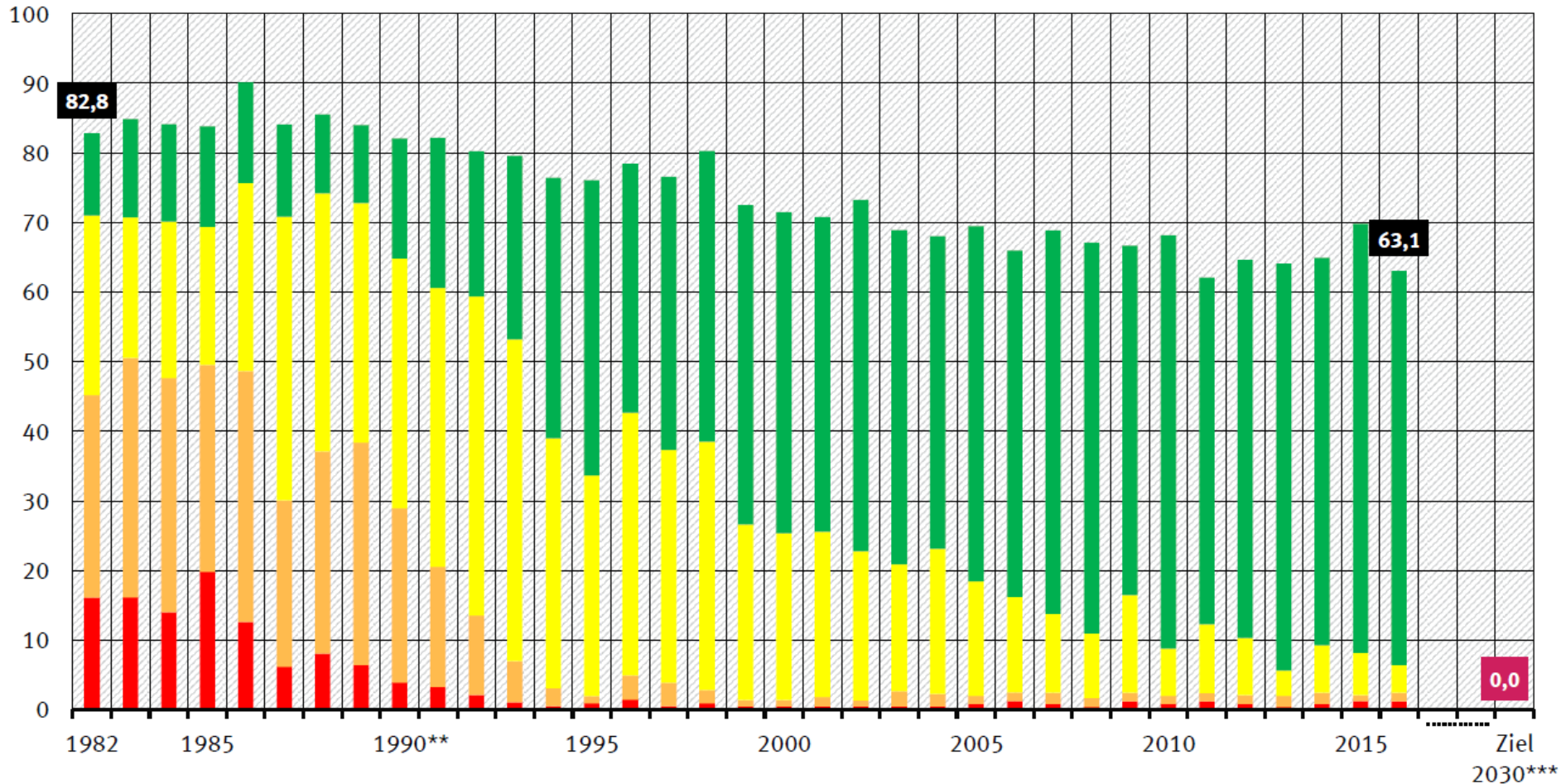
Objectives of P-removal

- Prevention of the eutrophication of water bodies especially in:
 - ➔ Lakes, dammed or tide influenced water bodies, Northern Sea of Baltic Sea
- Factor of minimum for the growth of plants: **N : P = 7:1**.
- Classification of lakes:
 - ➔ **Eutrophic**; nutrient rich lakes, in depth about 0-30%, O₂-saturation, only 2 m sight in depth, high production of algae, **tot. P~ 45 - 85 µg/l**
- River according the German Surface water ordinance
 - ➔ 50 µg P/l large rivers in the middle mountain; **inlets to the Baltic Sea**
 - ➔ 50/100 µg P/l all other rivers
- Limiting value of WWTP in Germany according wastewater ordinance
 - ➔ 10,000 - < 100,000 PE: 2.0 mg P/l
 - ➔ > 100,000 PE: 1.0 mg P/l
 - ➔ Special regional restrictions (e.g. Helcom agreement)!

Situation in Germany

Measurement points, which exceed the orientating value of total Phosphor

percentage



■ Quality class IV
 ■ Quality class III - IV
 ■ Quality class III
 ■ Quality class II- II

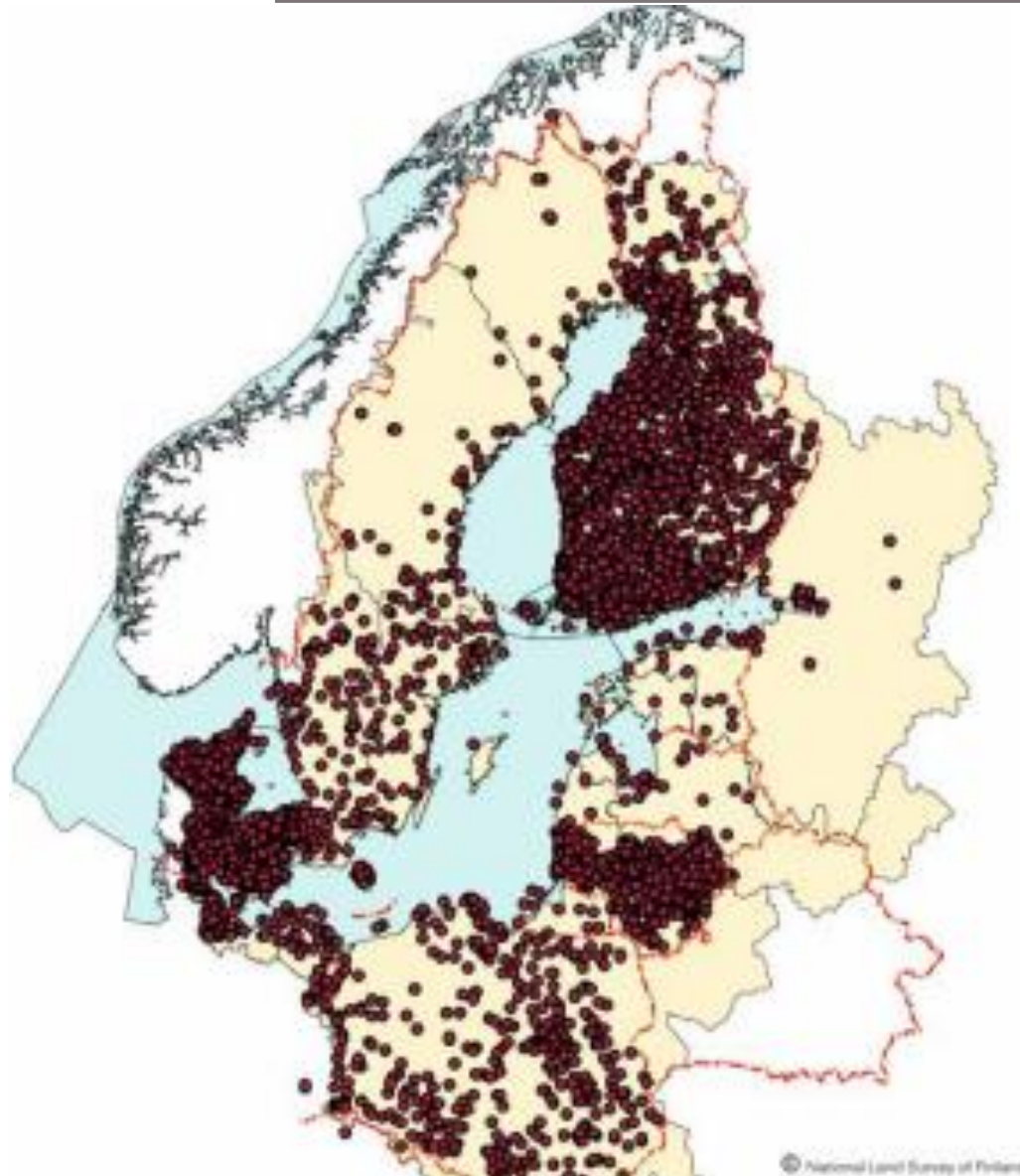
Objective: **Quality class II**

Some facts of the Baltic Sea

Baltic Sea facts

- A shallow (Ø depth: 53 m) and almost landlocked sea
- One of the largest bodies of brackish water in the world
 - ➔ Salinity 2 -20 ‰
- Unique and fragile ecosystems with naturally low numbers of species
- Highly sensitive to all kinds of pollution
- Surrounded by many large cities and regions with intensive agriculture and industry
 - ➔ 85,6 mio. people; Neighbouring countries: 9
 - ➔ Catchment: 1,7 mio. km²
- Some of the busiest shipping lanes in the world

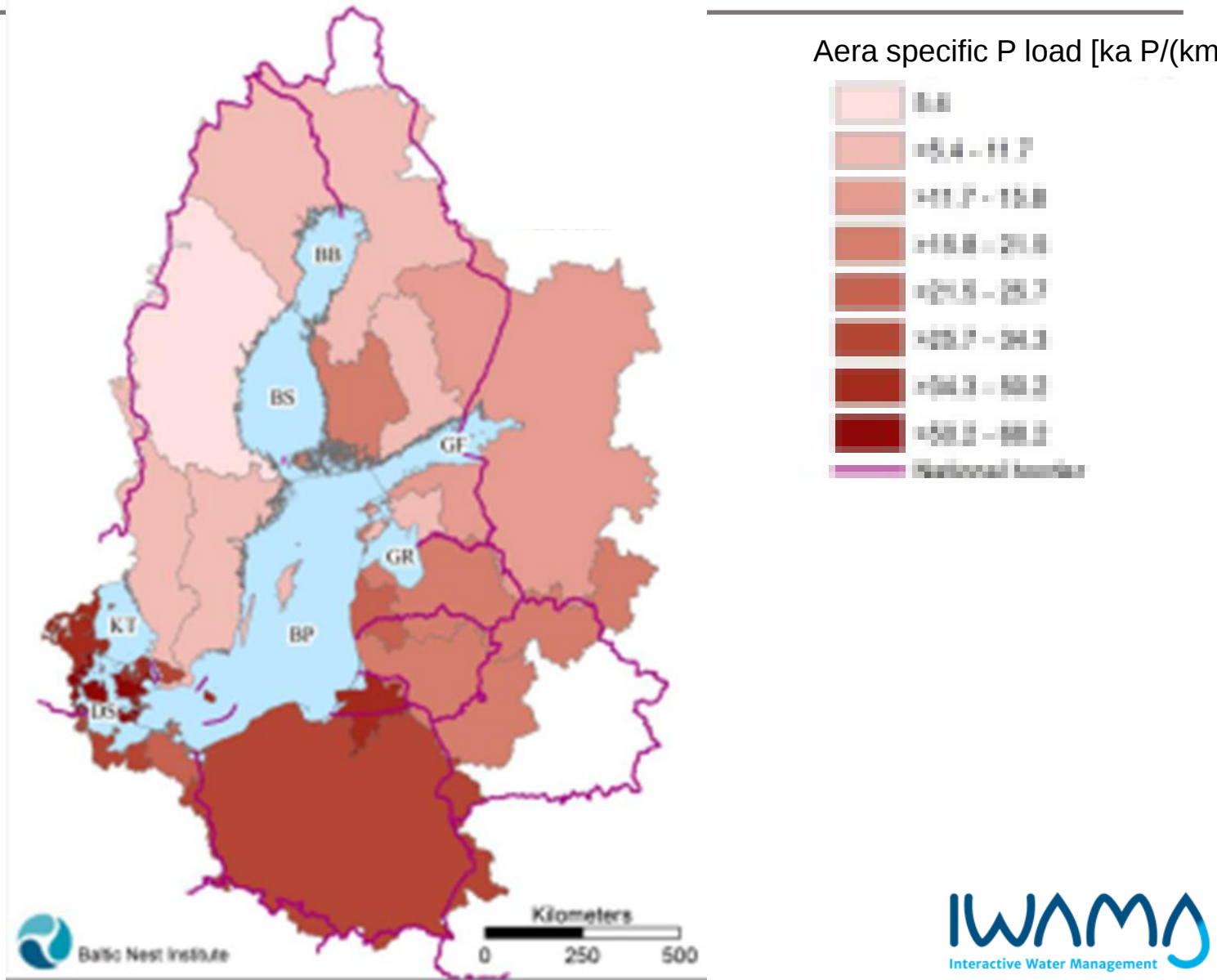
Treatment plant >10,000 PE (Left); Land cover, catchment area and sub-basins (right) of the Baltic Sea



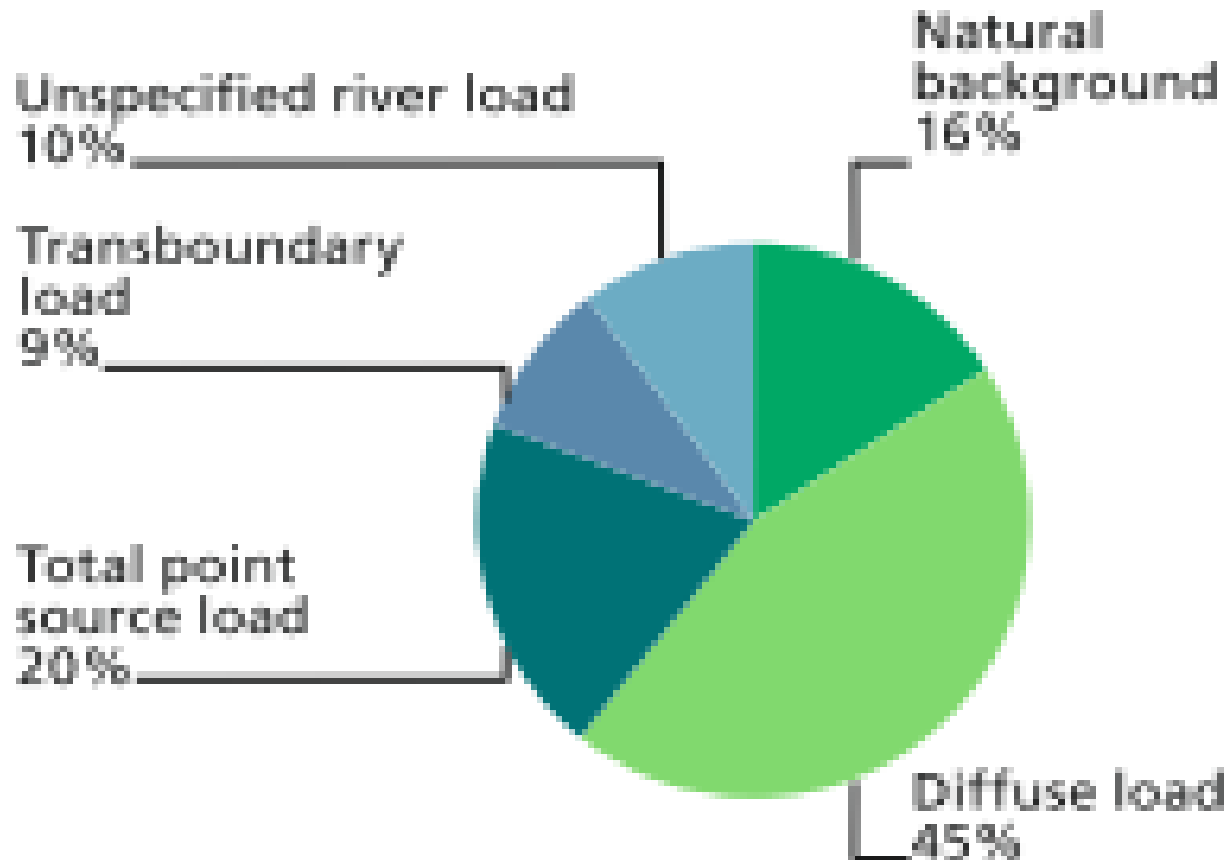
Source: Helcom

P-Emission into the baltic sea (2006)

Area-specific riverine phosphorus inputs ($\text{kg}/\text{km}^2 \cdot \text{a}$)



Proportion of different sources (in %) contributing to the phosphorus inputs into the Baltic Sea in 2006.



a) Total waterborne phosphorus 28,370 t

Source: Helcom

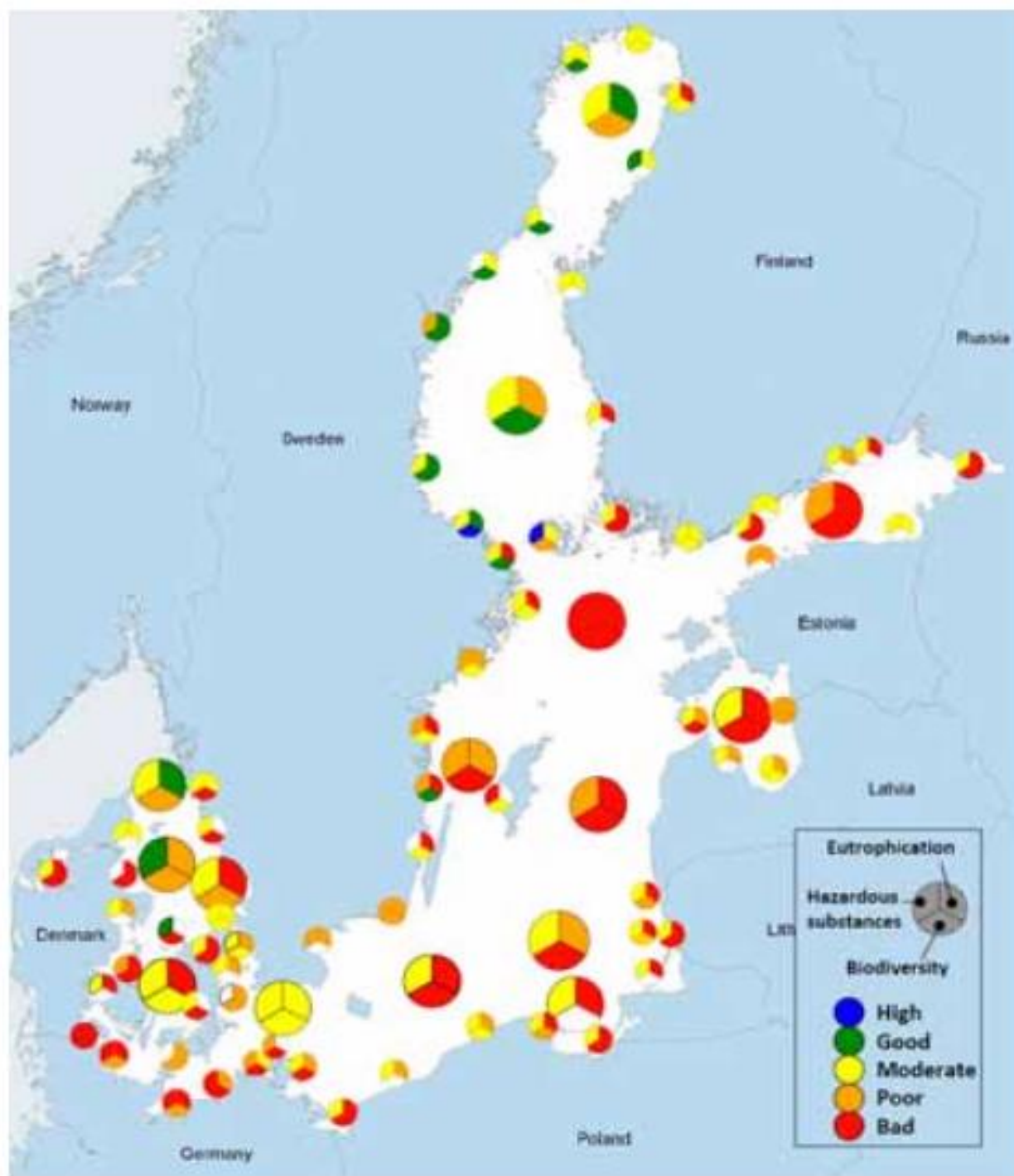


12 Taivallahti Bay, Helsinki, July 2013

Rügen, July 2013



Status of the Baltic Sea



Origin and estimation of the specific P-load in municipal sewage [in g P / (E·d)]

Origin of phosphorus	1975	2015
Metabolism products	1.6	1.6
Washing powder for textiles	2.25	0
Household detergents/washing-up liquid	0.75	0.24
Other P-sources	0.4	0.05
Sum, related to PE	5.0	1.8
Industry/trade/settlements (as PE)	1.2	1.2
Average Concentration in Germany (mg/l)	14	5.6

Only ortho-phosphate (PO_4^-) (68% of the inlet) can be removed from the water by precipitation and following particle separation!

Upcoming fraction of organic solved and none solved phosphorus
phosphonate, hardly degradable organic P-compounds from
textile industry and from heat-/power-production

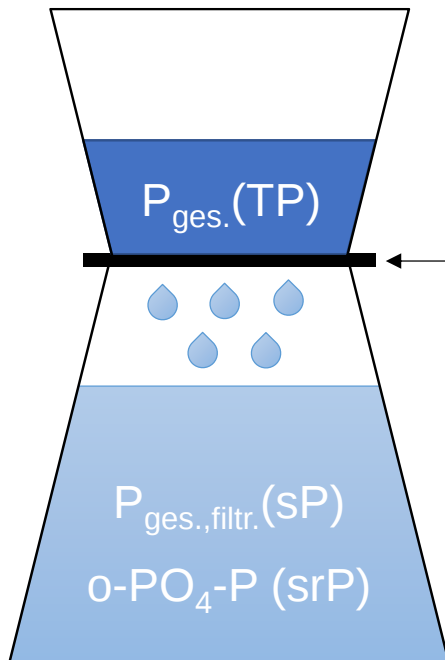
Fractions of Phosphorus in the sewage

Calculation snrP

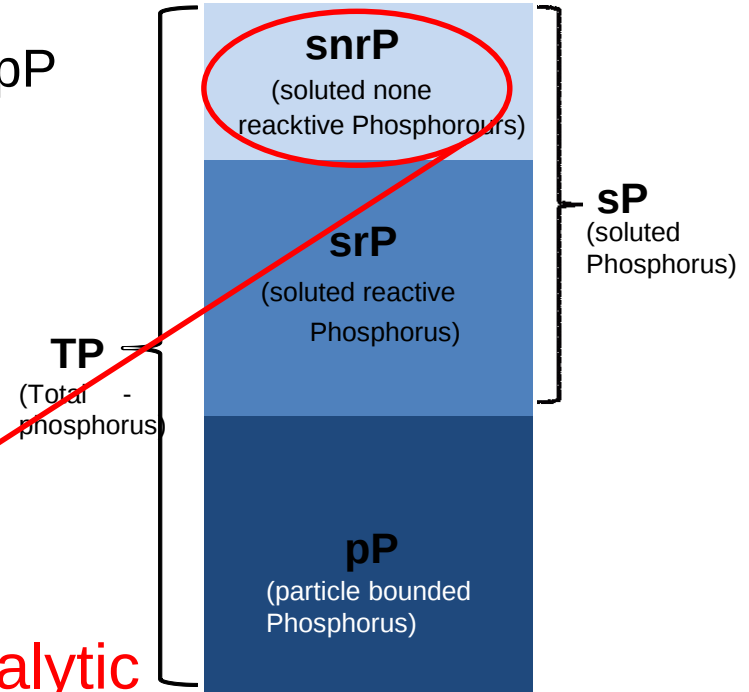
$$pP = TP - sP$$

$$snrP = TP - srP - pP$$

$$snrP = sP - srP$$



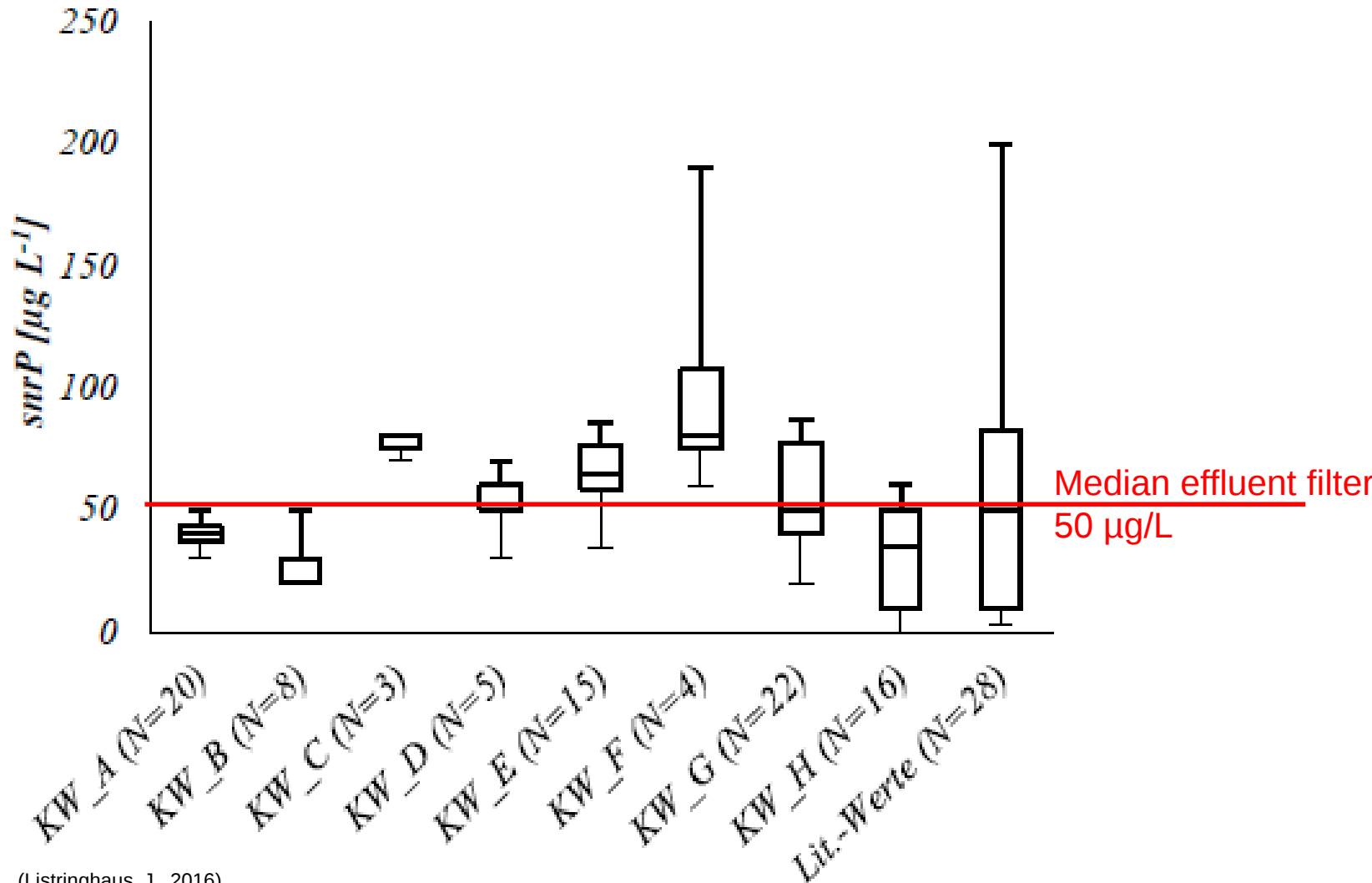
Membranfilter
(0,45 µm)



Relation to the analytic

→ P-compounds not reactive with flocculants

snrP effluent values of biofilter Comparison to literature



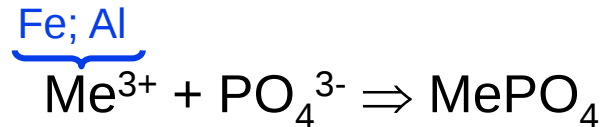
(Listringhaus, J., 2016)

Chemical and biological aspects

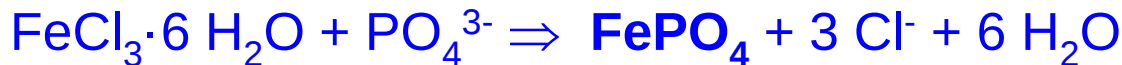
Chemical phosphorus removal

Coagulant and reactions

Applying metal salts:



Reaction with the coagulant (example iron chloride (FeCl_3)):

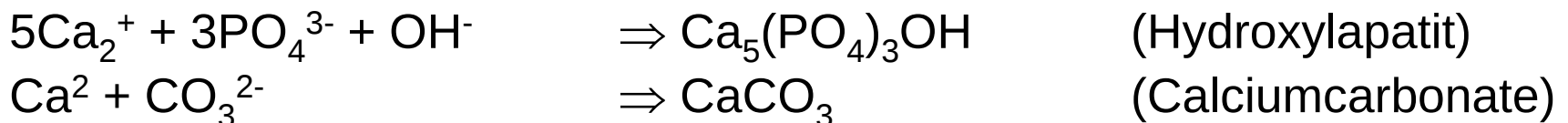


Unavoidable by-reaction (Building of (Hydroxide flocs):



- ➔ Building of hydroxides requires always an overdosage of coagulant → β value
- ➔ And an additional sludge production

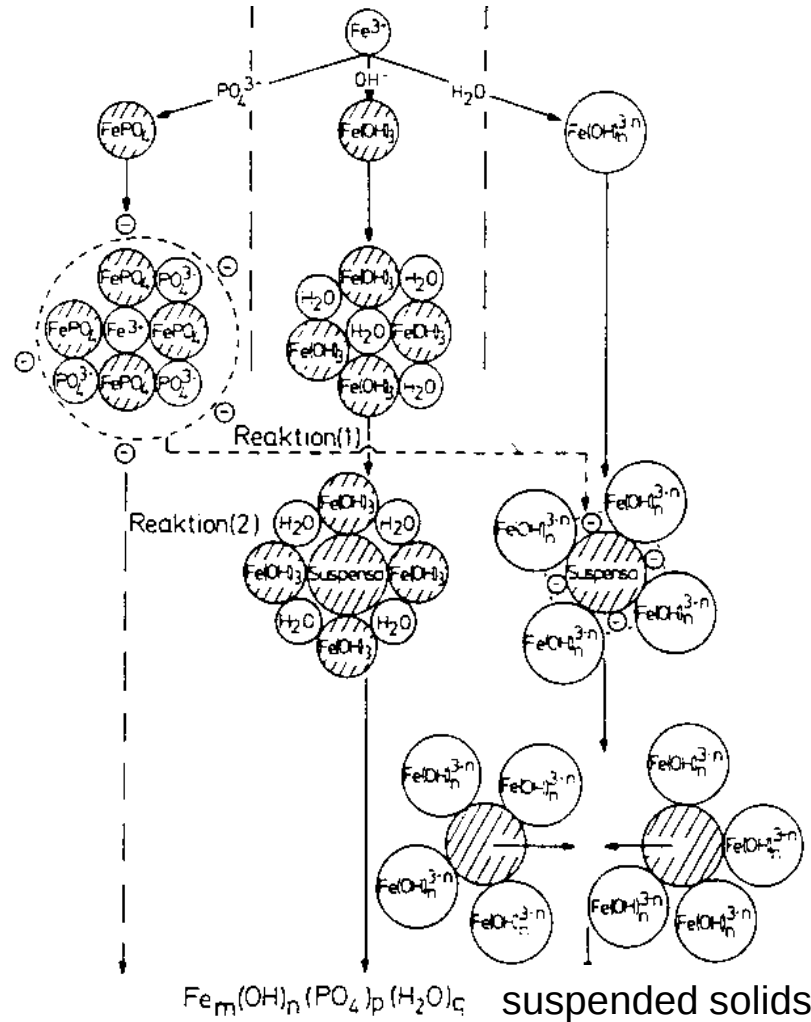
Applying calcium (2 different precipitation products):

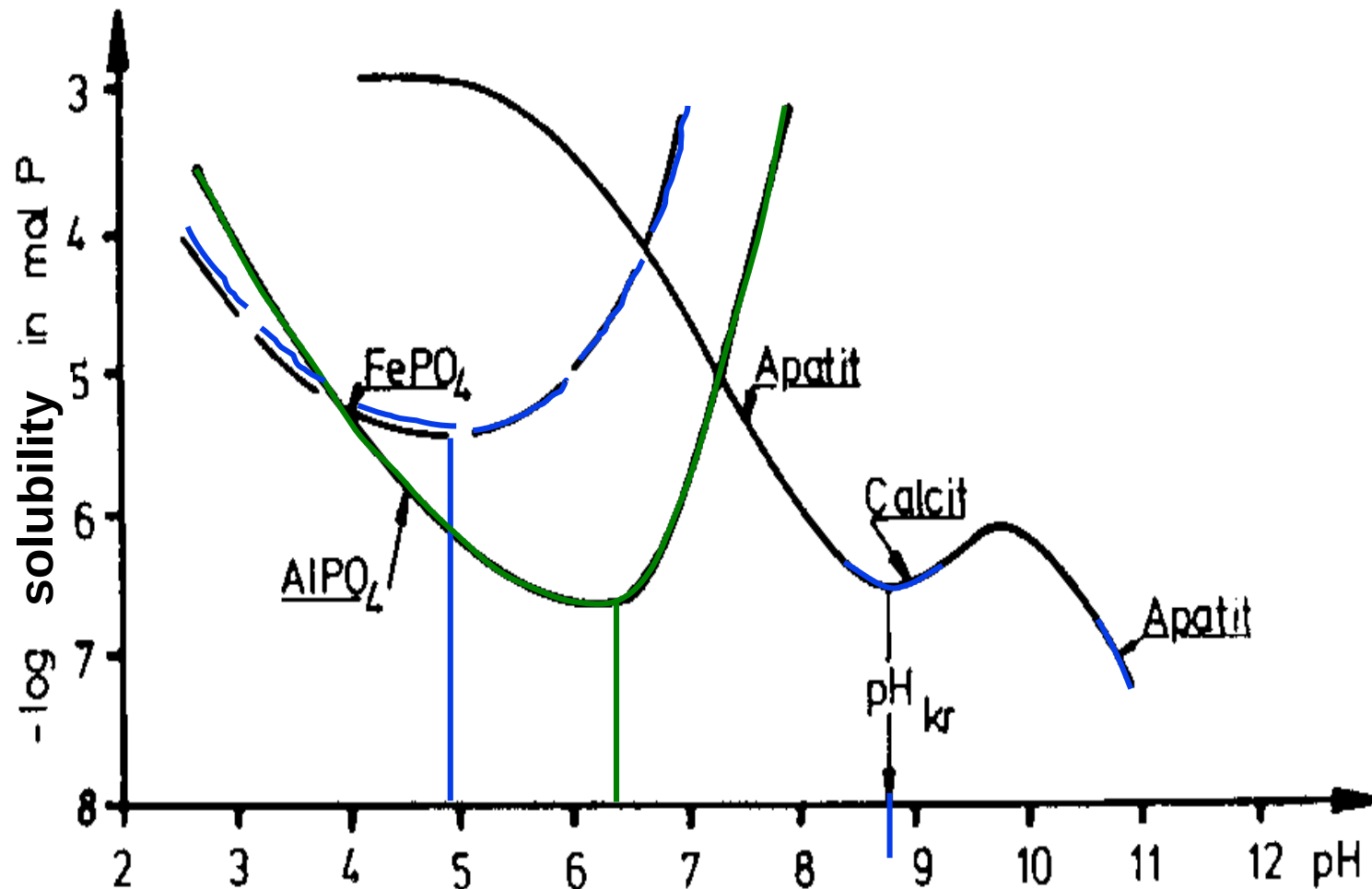


Flocculation process

Precipitation Co-flocculation ; Flocculation

2 Dosage of coagulant





Calculation of the amount of the precipitant

β -value: relative amount of precipitant with the unit [mol Me/mol $X_{P, Prec}$]

$$\beta_{prec} = \frac{X_{Me} / AM_{Me}}{X_{P, Fäll} / AM_P} \quad [mol Me / mol P]$$

$\beta = 1,0$ stoichiometric dosage

$\beta = 1,5$ 50 % over stoichiometric dosage

with:

X_{Me} necessary amount of precipitant (metal) (mg Me/l sewage),

$X_{P, Fäll}$ phosphorus to be precipitated (mg P/l sewage),

$$X_{P, prec} = C_{P, In} - C_{P, out} - X_{P, BM} - X_{P, DN} - X_{P, BioP}$$

AM_{Me} atomic mass of metal (mg/mmol),

AM_P atomic mass of phosphors (mg/mmol)

and the atomic mass AM: (P) Phosphor: 31; (Fe) iron 56 ; (Al) aluminium 27

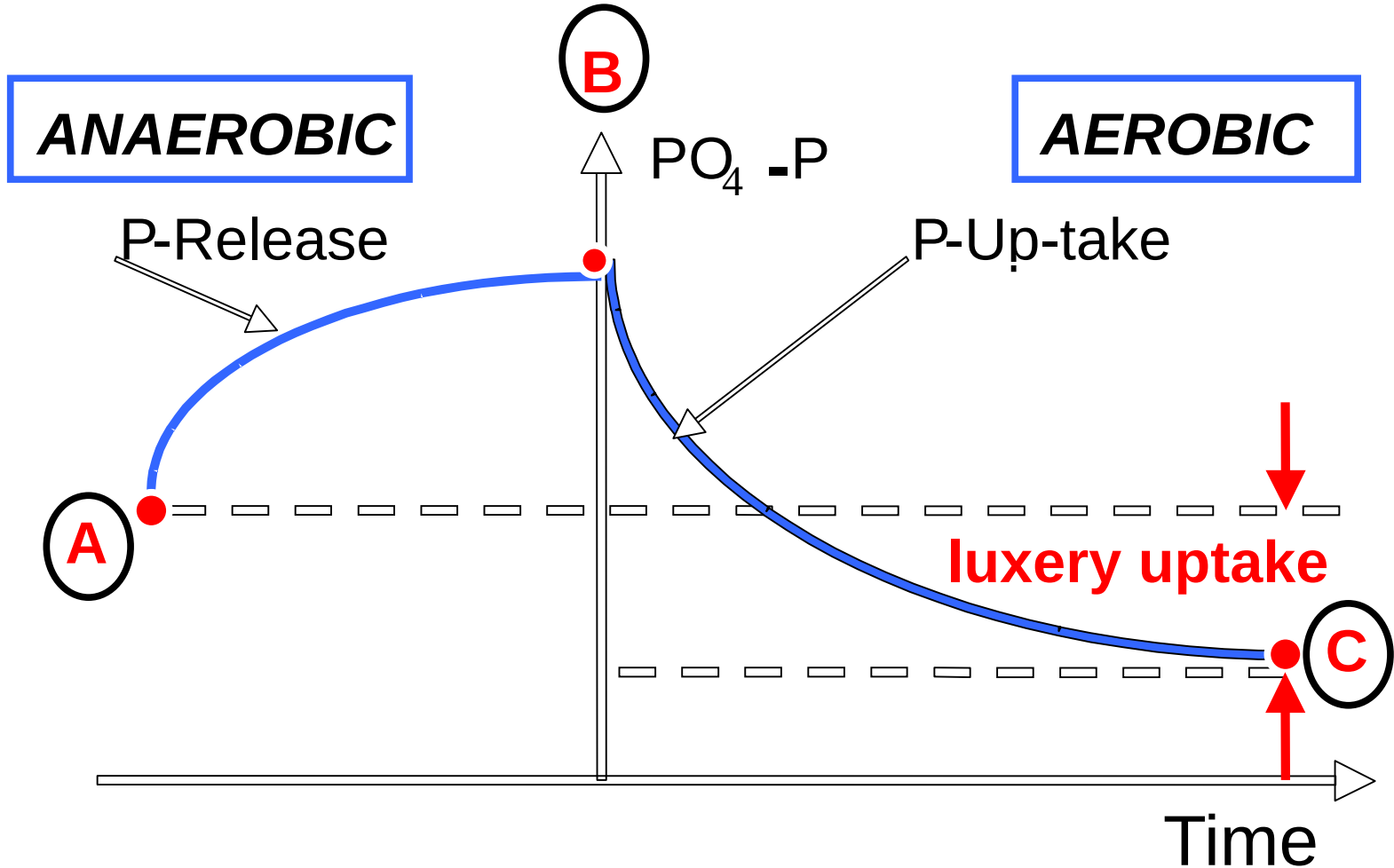
According the kind of bounded P in the sludge:

X_{PBM} = 1,0 % (nitrification, COD Elimination)

X_{PBDN} = 0,5 % (denitrification)

X_{PBioP} = 1,0 % (anaerobic tank)

Principle of the enhanced biological P-elimination



Advantages:

- ➔ No additional agents (flocculants) necessary
- ➔ Decrease of salts in the effluent of a WWTP
- ➔ Lower additional sludge production
- ➔ No additional heavy metals in the sludge
- ➔ No disturbances possible for the nitrification process

Disadvantages

- ➔ Higher investment cost for the construction of the anaerobic tank
- ➔ Sometimes instable process, so often additional chemical dosage is required
- ➔ In winter time higher tendency of bulking sludge
- ➔ High content of volatile fatty acids ($> 100 \text{ mg/l}$)
- ➔ Sufficient O_2 -input into the activated tank to realize a high P-uptake

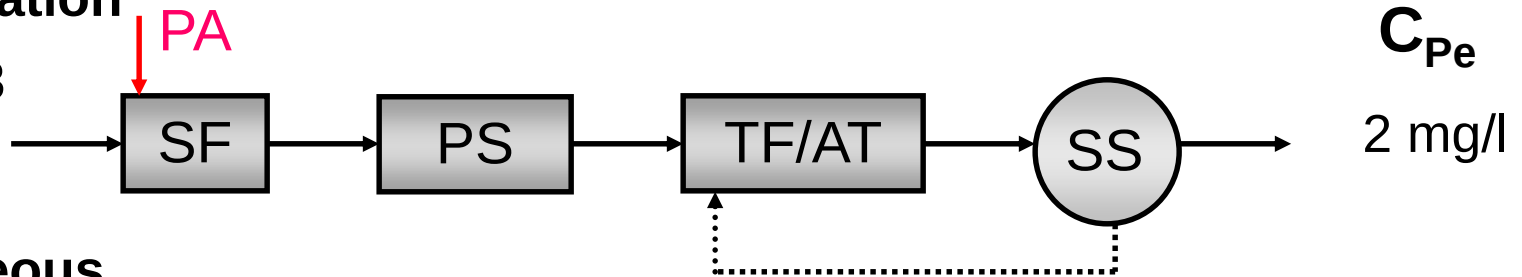
Process technology of phosphorous removal

- Chemical, biological -

Processes of the chemical P-elimination

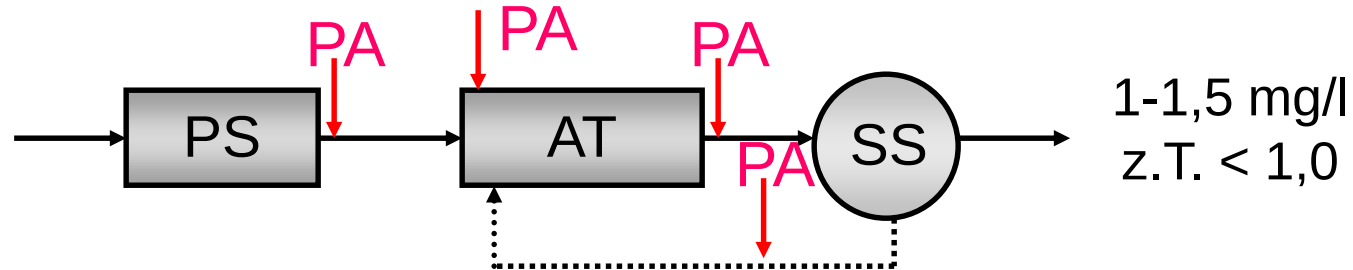
Pre-precipitation

$\beta = 2-3$



Simultaneous precipitation

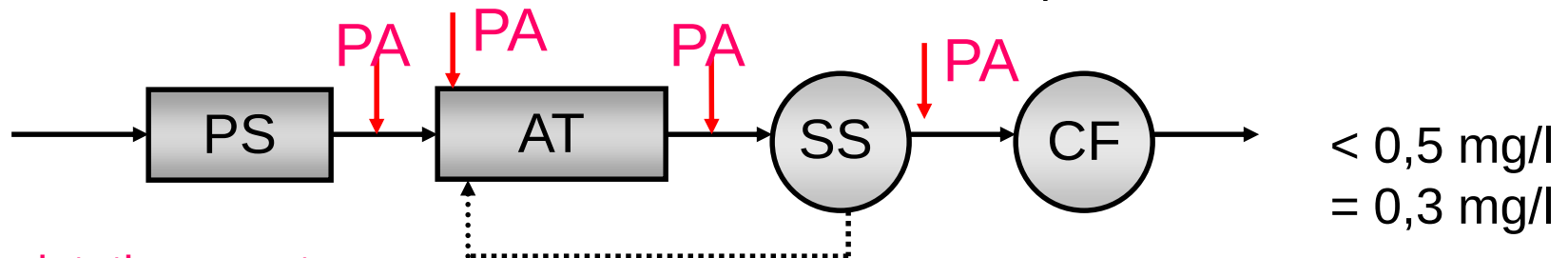
$\beta = 1,2 - 1,5$



Combined precipitation

$\beta = 1,2 - 1,5$

$\beta = 2 - 2,5$



PA: precipitation agent

Dosage of the precipitation agent

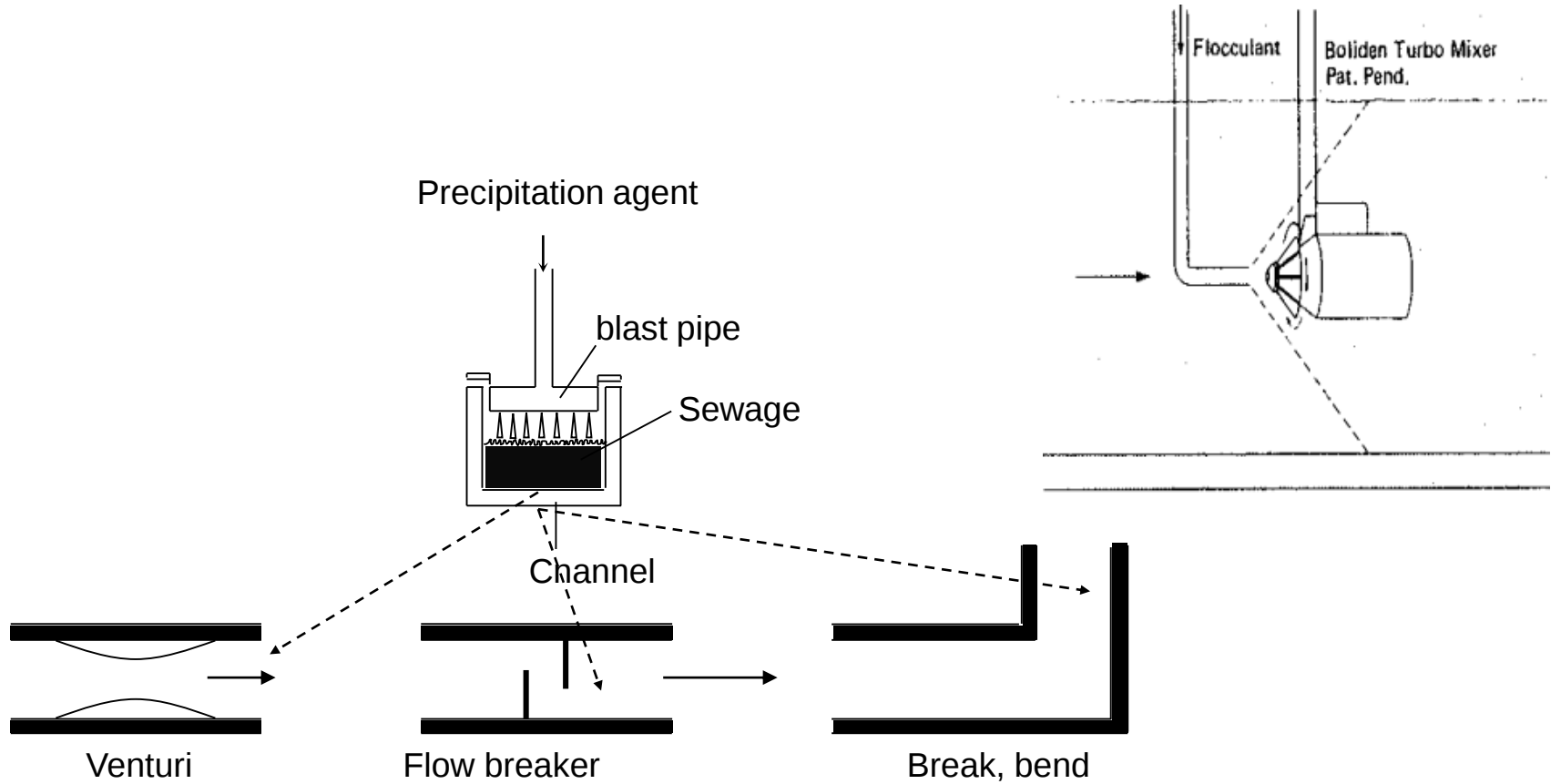
- at **locations with** by flow **increase turbulence**
- in short distance of the storage tank
- hydraulic direct jump
 - ➔ Increase of the cross section,
 - ➔ absorption chamber of weirs,
 - ➔ Drop construction
 - ➔ junction of part flows
- Installation of mixer with a high frequency
- Installation of screens and static mixers
- Archimedes screw (e.g. filtration)
- Pressure side turbulence of pumps
- Aerated flow section



Protection pipe

(Fe²⁺)

Example of good location for the dosage of precipitations agents



Examples of location for dosage



Bad solution

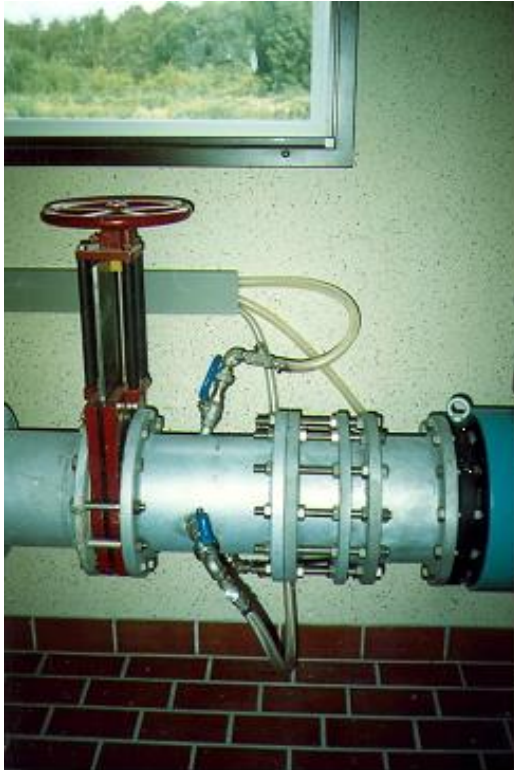
- bad mixing of the flocculant
- durch Randlege kaum B nearly no contact to the complete water surface



good solution

- Dosage into the change of flow direction
- but better would be over the whole cross-section

Examples of location for dosage



Good mixing conditions

- special injections form different side of the pipe
- Against the flow direction



- also good mixing conditions
- but danger of corrosion

Storage of the flocculation agent

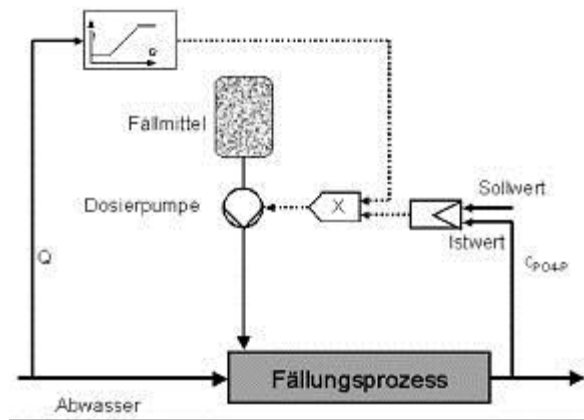


- Sealed filling place
- Separated catchment of losses of the flocculation agent



Control concepts

Process	Size of WWTP	Amount of flocculation agent
Control according pre-seted time	Not recommendable (only small WWTP)	--
Control according a flow graph	Small WWTP	-
Control in relation to the flow	Mid sized WTTP	+/-
Control according to the load in the inlet	Large WWTP	++
Control of $C_{PO4-P, \text{Effluent}}$	Large WWTP	++





Example

WWTP Lübeck-Priwall

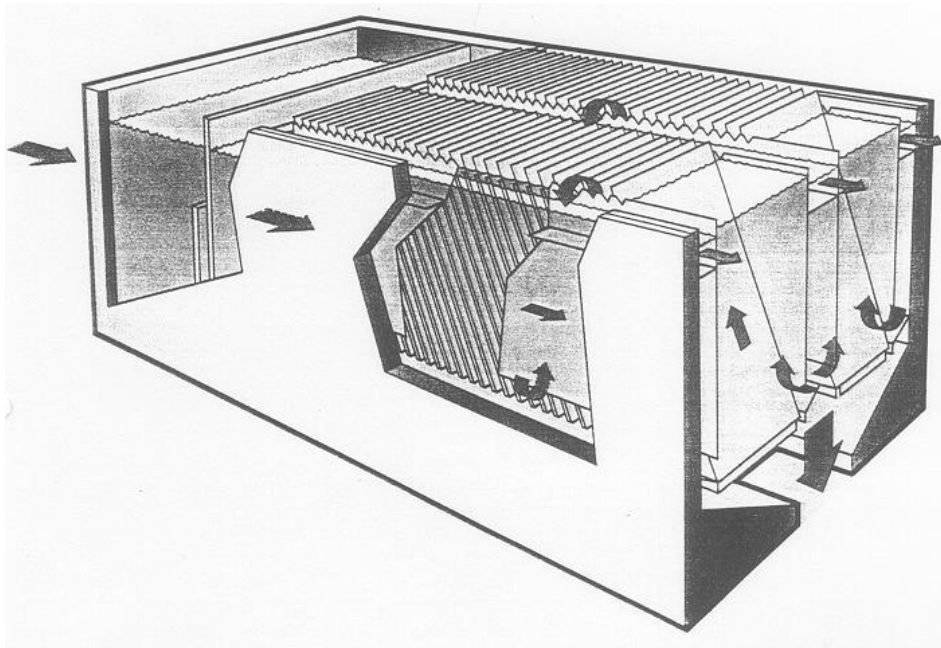


Advanced Phosphorous removal technologies

Processes for advanced P-removal

- First step:
Residual precipitation and flocculation of Phosphorous
 - ➔ Dosage of agent and flocculation tank is required
- Post precipitation and separation of the flocs with:
 - ➔ Settling tank, Lamella separator, flotation
 - ➔ Micro-sieve
 - ➔ Cloth filter
 - ➔ **Coagulation filtration**
 - Down flow deep bed filter
 - | Discontinuously back washed
 - | Continuously
- Membrane-filtration (post positioned)
- Biomembrane-reactor

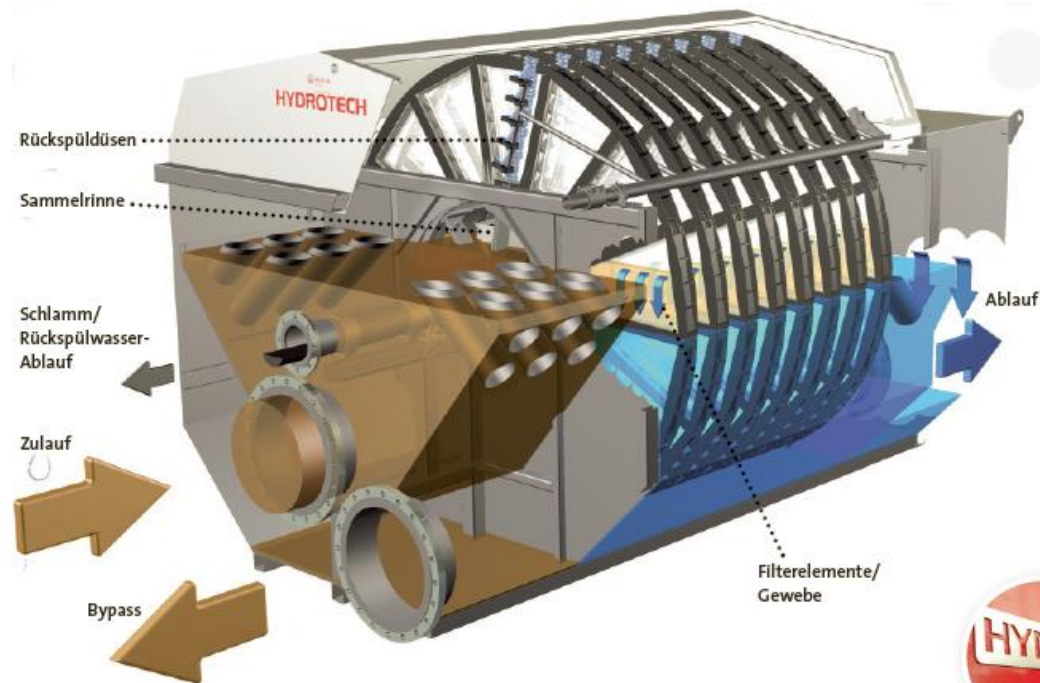
Operational results of lamella separator Post-precipitation (5 plants)



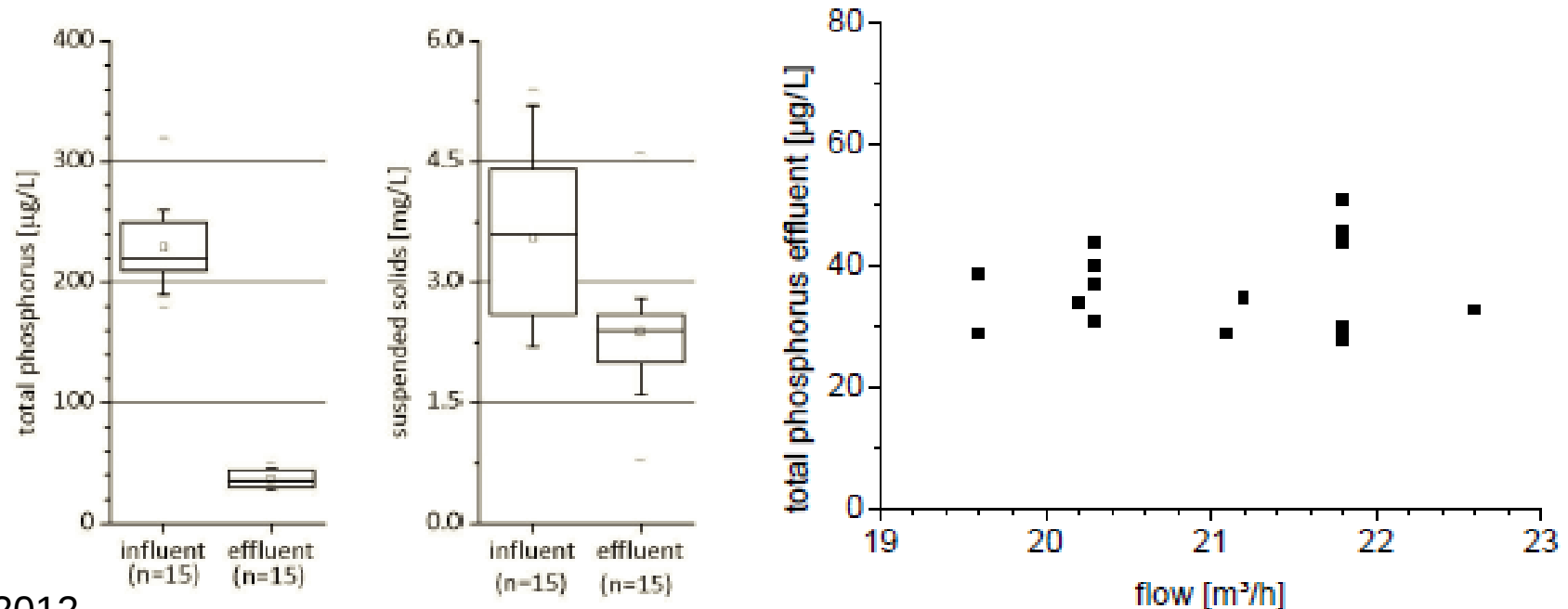
	AFS ₀ ¹⁾	AFS _e	η	gesP ₀	gesP _e	η	CSB ₀	CSB _e	η
lamella	[mg/L]	[mg/L]	[%]	[mg/L]	[mg/L]	[%]	[mg/L]	[mg/L]	[%]
Average	9,5	3,0	73	4,46	0,62	86	38	36	29
Minimum	13,0	0,3	-	2,14	0,21	75	34	24	29
Maximum	6,6	6	-	5,81	1,44	92	42	90	30

Modern Micro-sieve (Hydrotech)

- closed meshed, monofile texture (width 10 – 100 μm)
- System is only damed to 2/3 of the diameter
- Filter velocity $v_{F,TW} \approx 10,0 \text{ m/h} - 20,0 \text{ m/h}$
- Smaller particles than mash width removable (5% open space)
- Cleaning with high pressure injection
- Low energy costs by gravity flow
- TSS-removal, P-Elimination



- Pre-located two stage coagulation suitable to reached P effluent value < 0.08 mg/L.
 - ➔ E.g. with 2 mg/L PACl (Polyaluminiumchloride) and 0,6 mg/L FHM (flocculation agent) ➔ 0,06-0,065 mg/L

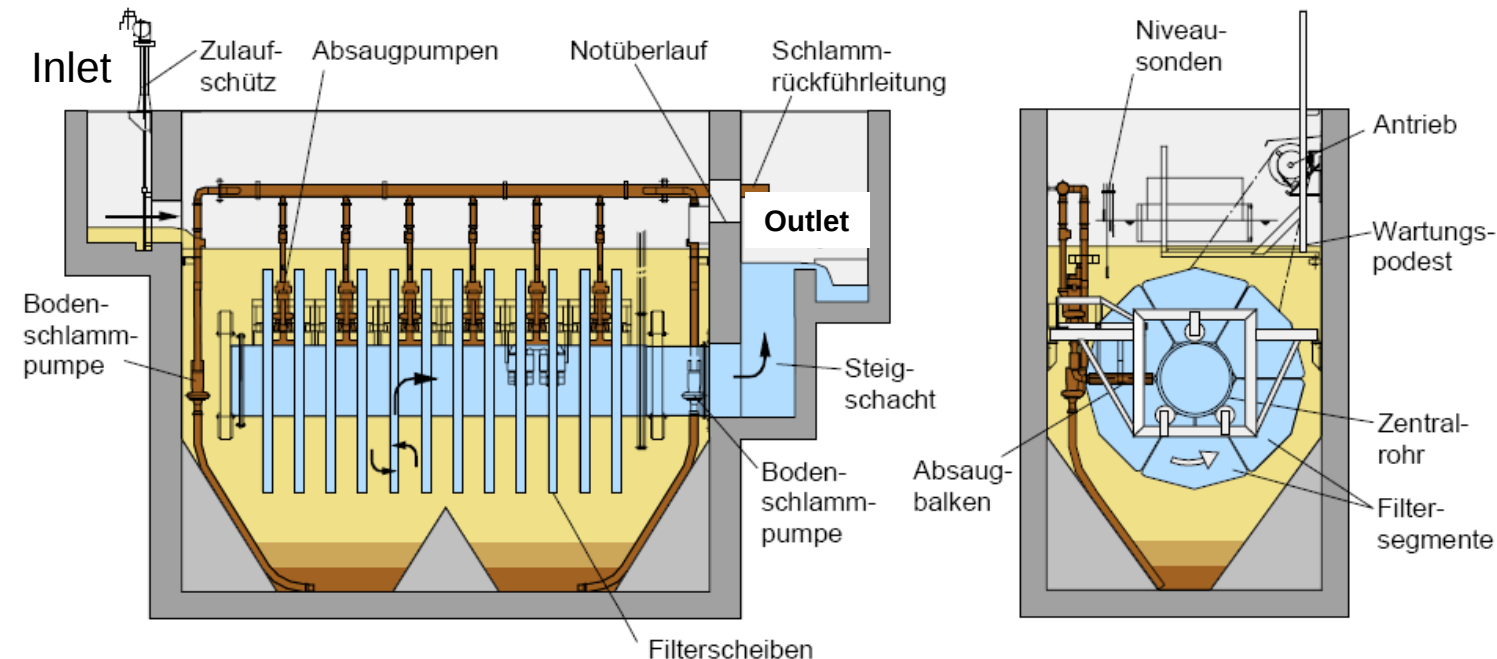


Lardon 2012

Cloth filtration

Disc-filter

- Formally small fibred needle felt cloth (width 10 – 10 μm)
today **Pile fabric**
- Filter velocity $v_{F,DW,RW} \approx 10,0 \text{ m/h}$
- Surface load of particles $< 400 \text{ g}/(\text{m}^2 \cdot \text{h})$; TSS_{in} max. 40 mg/l
- Suitable for TSS- and P-removal



Basic space of drum filter: $3,8 \text{ m}^2 A_F / \text{m}^2 A_{\text{base}}$

Cloth filtration

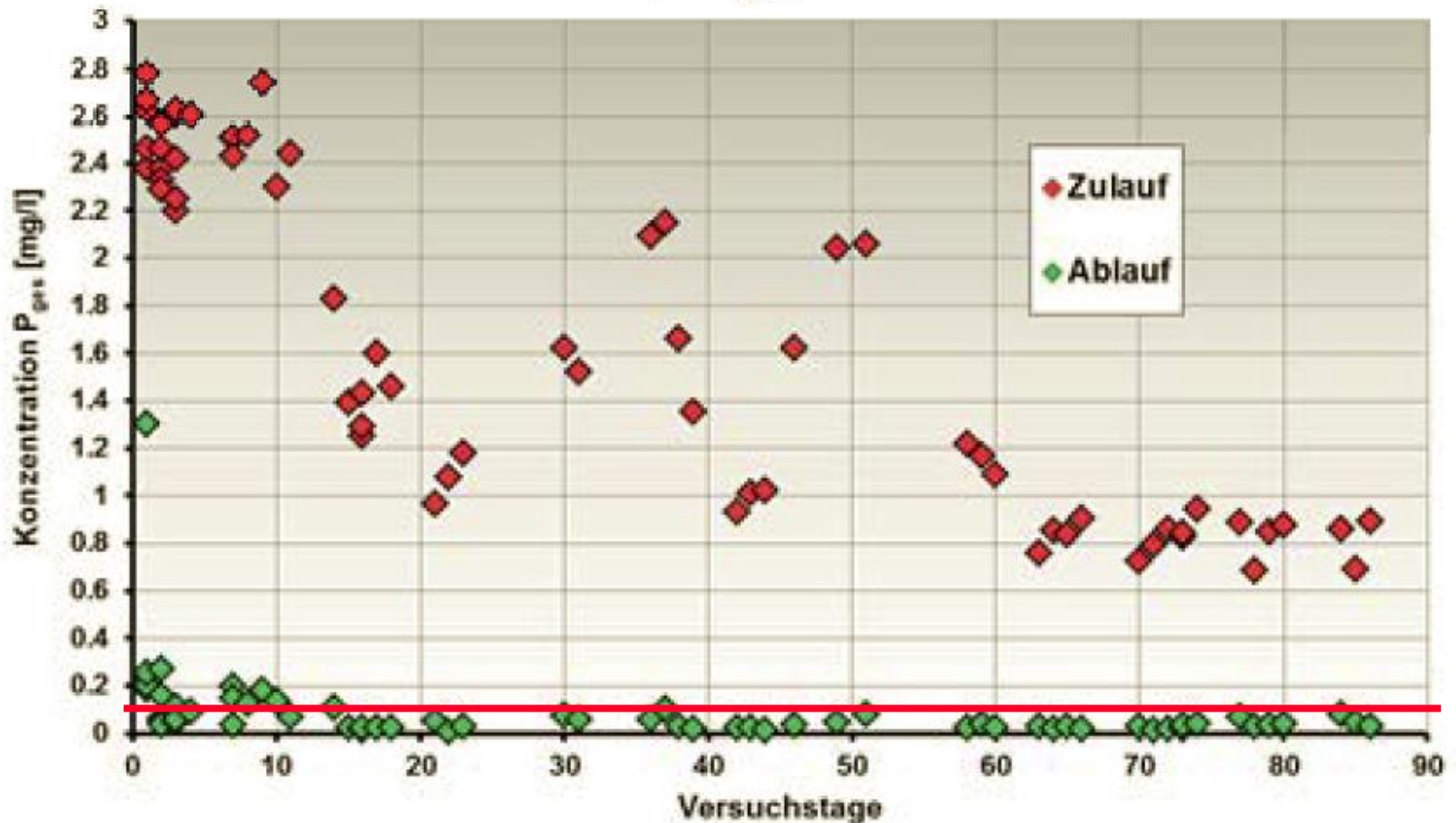
Disc-filter; Example WTP Oldenburg

- 14 x 60m² Disc filter
- Total filter surface 840 m²
- Maximum flow: 5,800 m³/h
- Today world wide largest disc cloth filter
- Built in 2006
- Effluent < 5 mg/l TSS
- Microplastic removal



P-Elimination with cloth filtration

Pole fabric coated

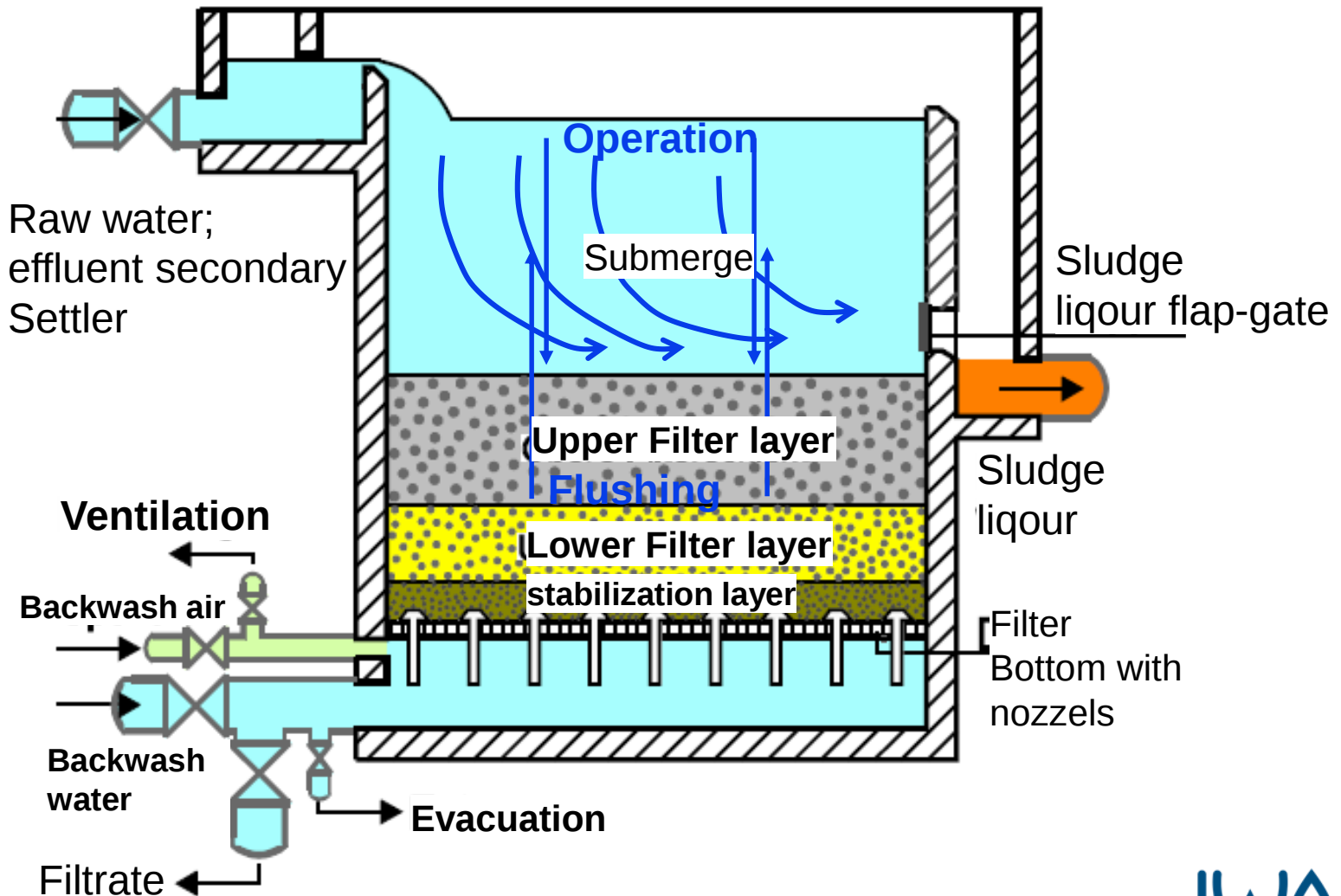


- Effluent < 0,1 mg/l P_{tot} with precipitation and pile fabric

Deep bed filter

Down-flow granular media filter

Intermediated backwashed by flap flushing

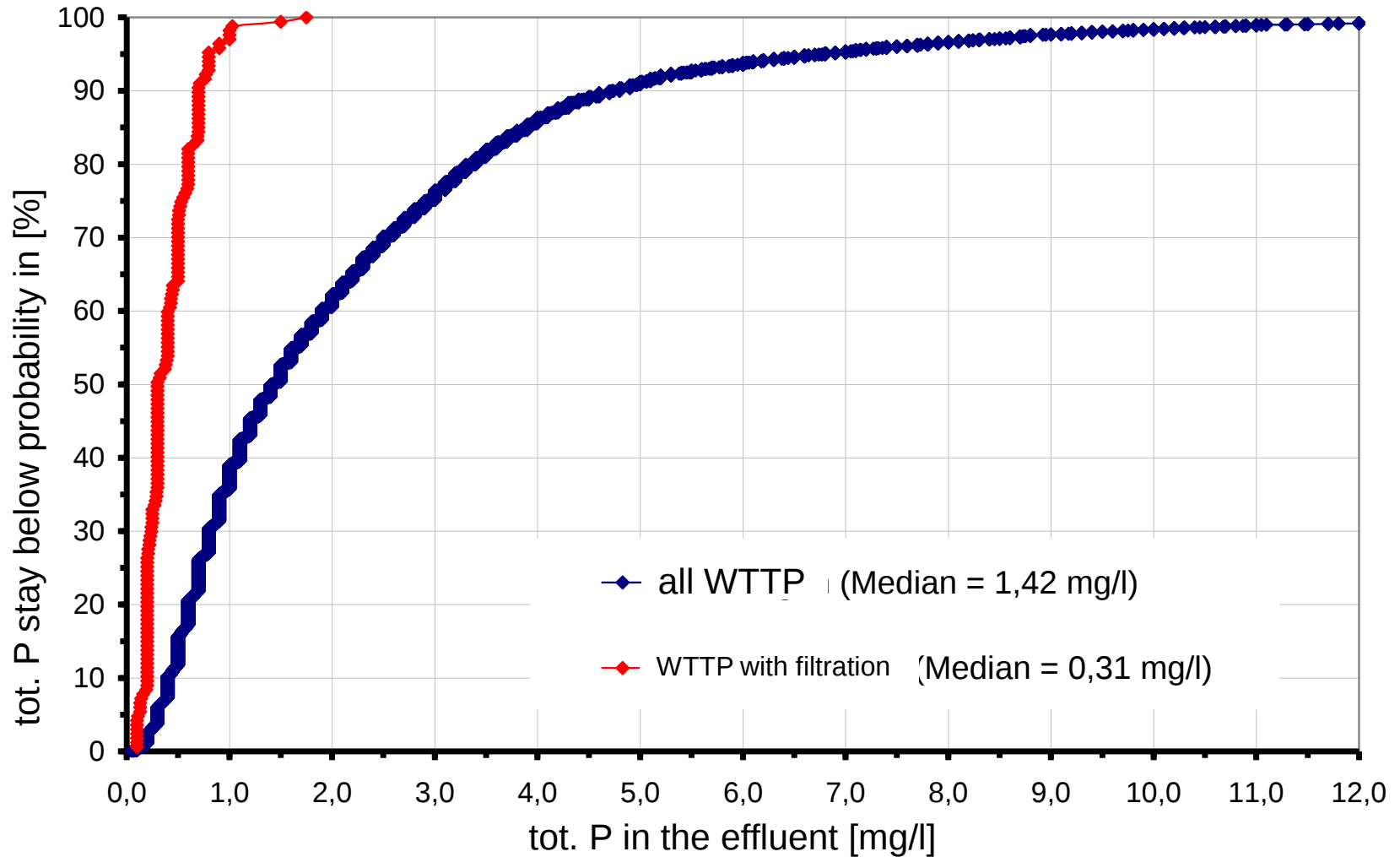


Down-flow granular media filter

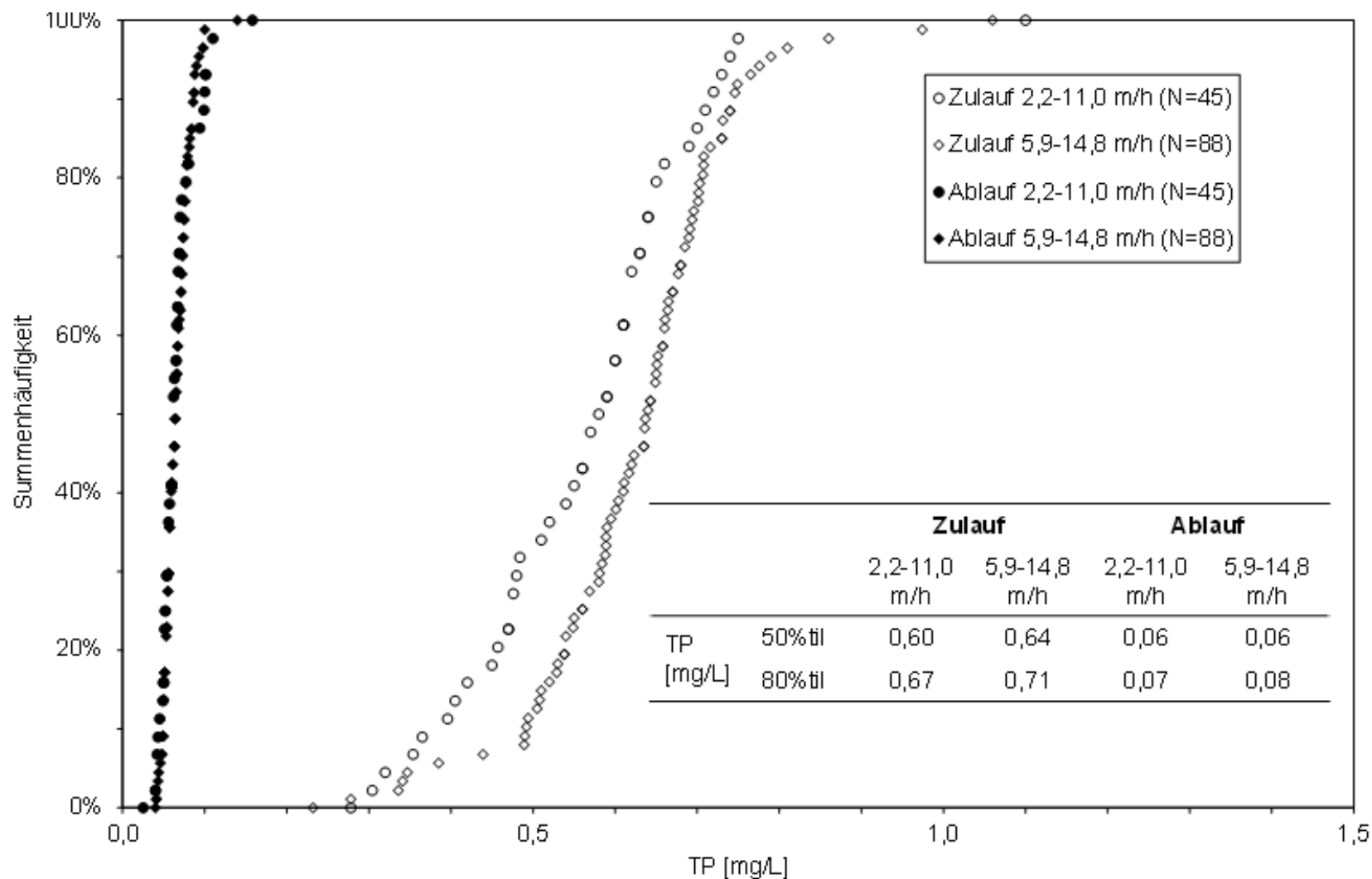
Intermediated backwashed by flap flushing



Operational results of coagulation Filtration

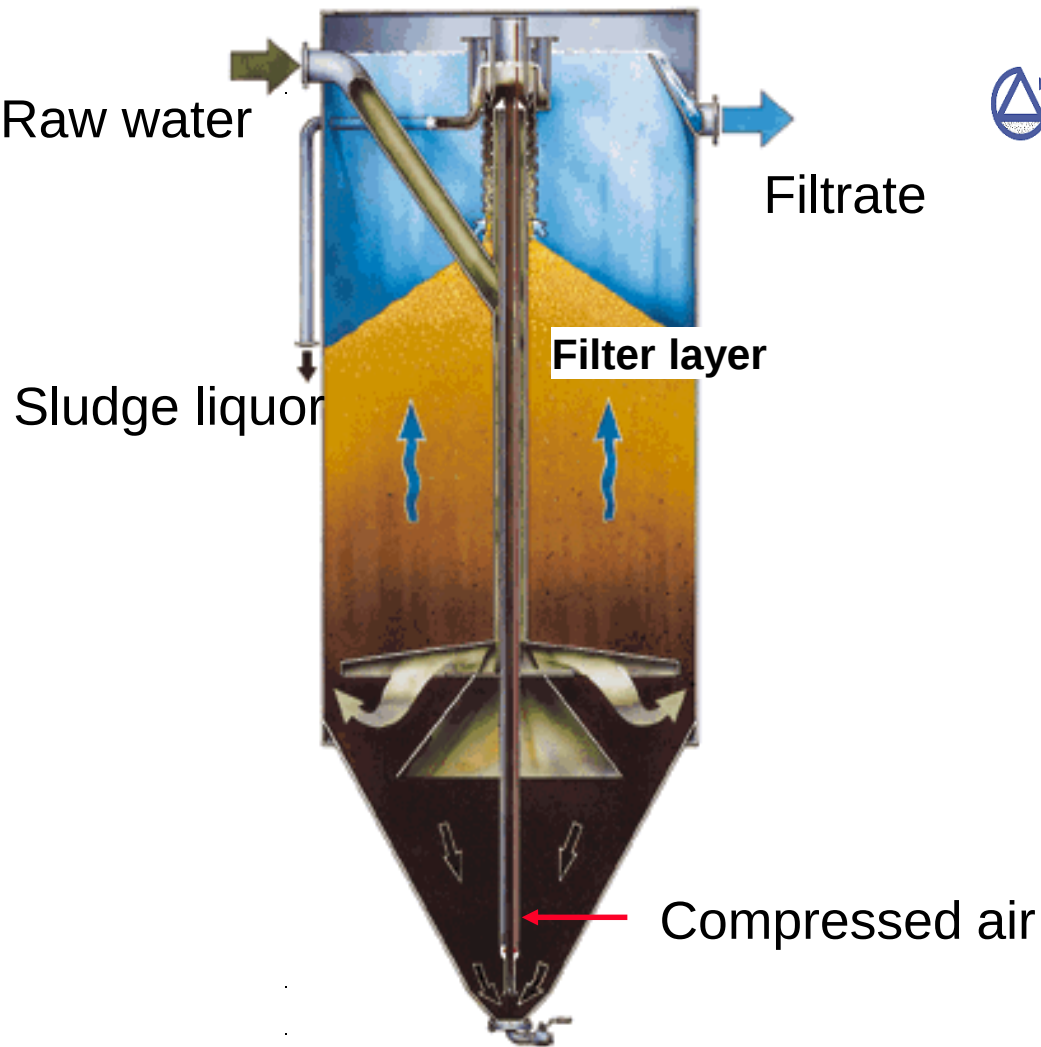


Stay below probability TP-Concentration in In- and outlet of a coagulation filter Dosage of Fe^{3+} and different v_F

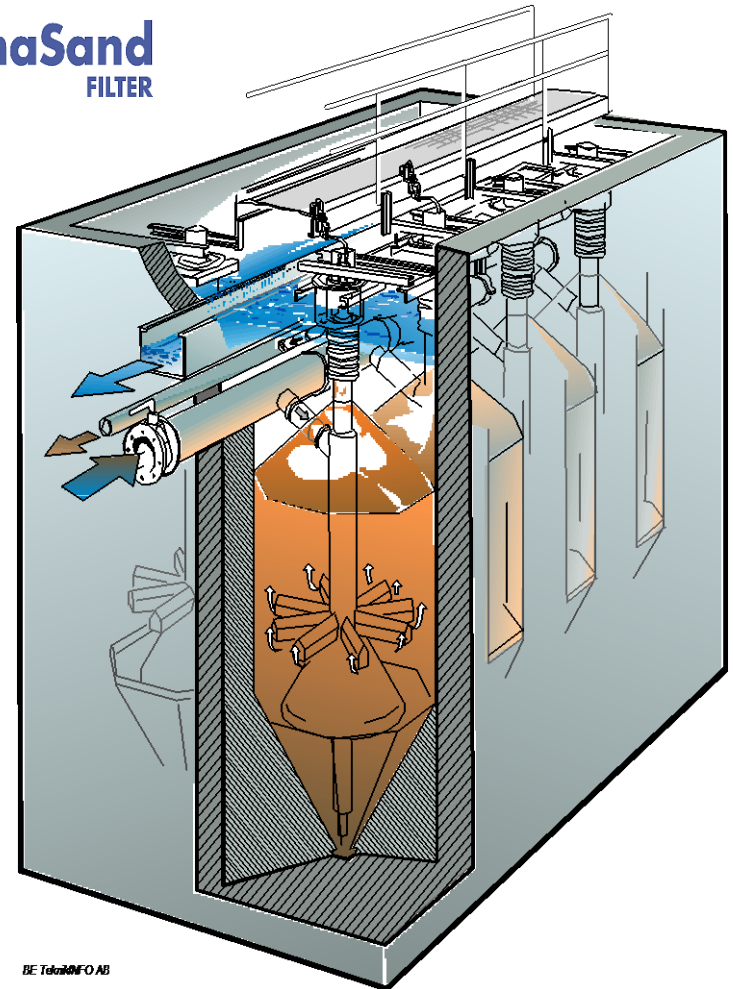


Continuously operating filter plants

Up-flow-Operation

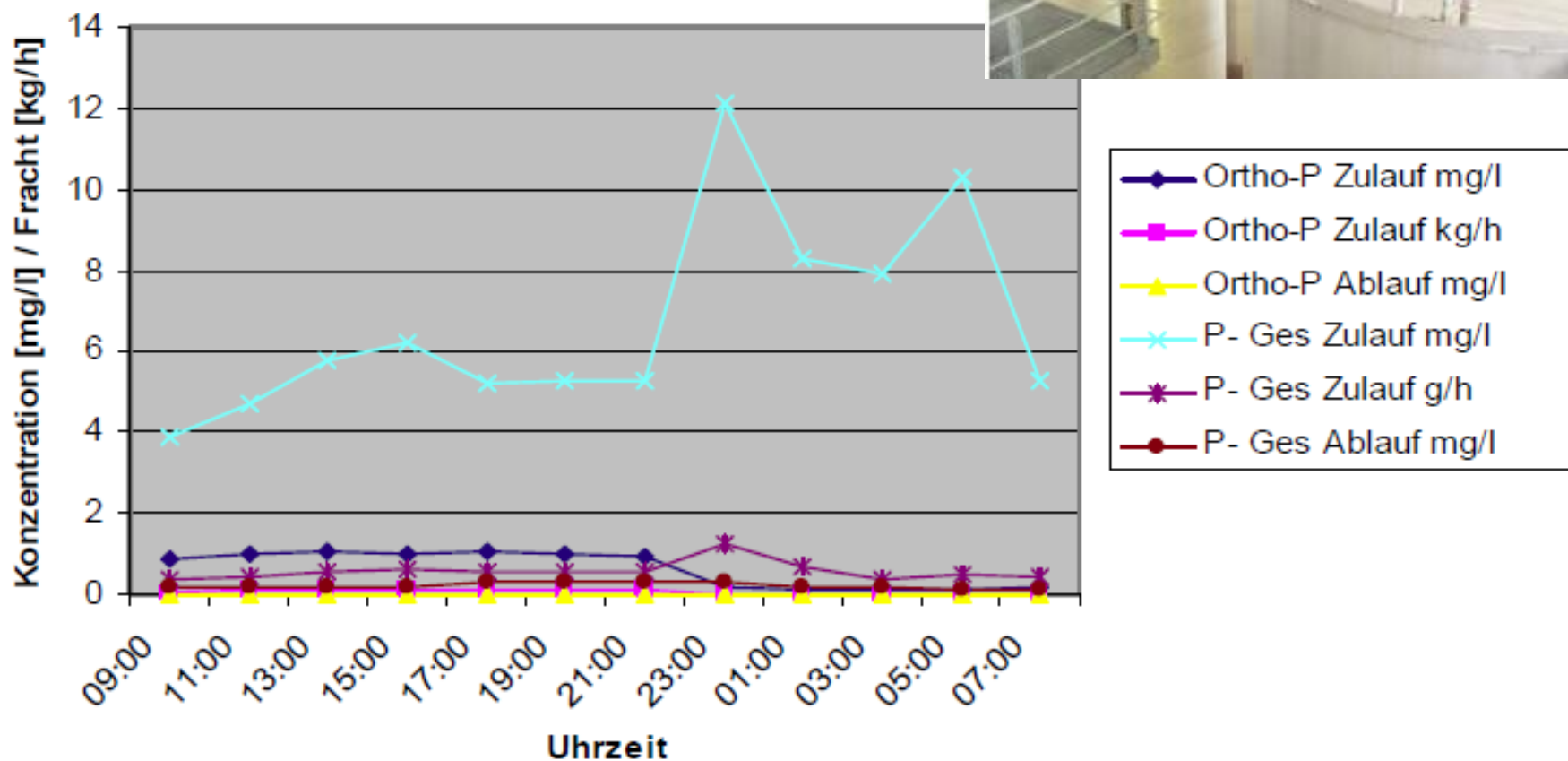


 **DynaSand**
FILTER



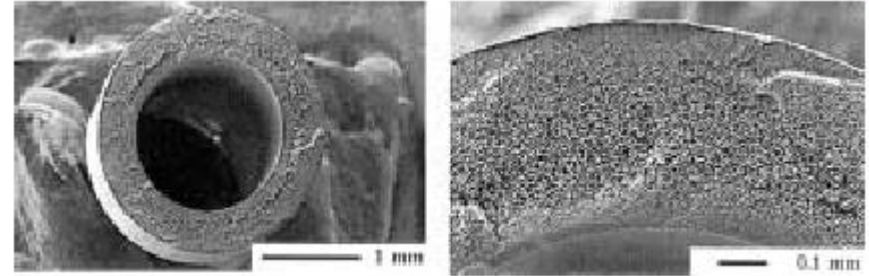
Phosphorous removal

Example WTP Ratzeburg



- Variants of modules

- ➔ pipe-, capillary-, Hollow fibre and surface module)
- ➔ Membrane material (organic - inorganic)
- ➔ Pore size from 0,01 to 0,2 μm
- ➔ Pre-located granulated filter necessary (removal of particles $> 500 \mu\text{m}$)



- Cleaning with gas ca. all 30 min.

- Flux-rate 41.6 l/(m²·h) with 1.3 bar starting pressure

- Energy consumption: 0.2 kWh/m³

- Results

- ➔ TP-Concentration from 0.2 - 1 mg/L down to $< 0,05 \text{ mg/L}$ in the effluent; monthly average **0.025 – 0.045 mg/L**
- ➔ β -value = 2 – 5; no coagulant agents

TP-Effluent of Biomembrane reactor

Germany

Belebungs

P_{tot} effluent values

sehr gering
gering

Zulauf

P_{ges} [mg/l]

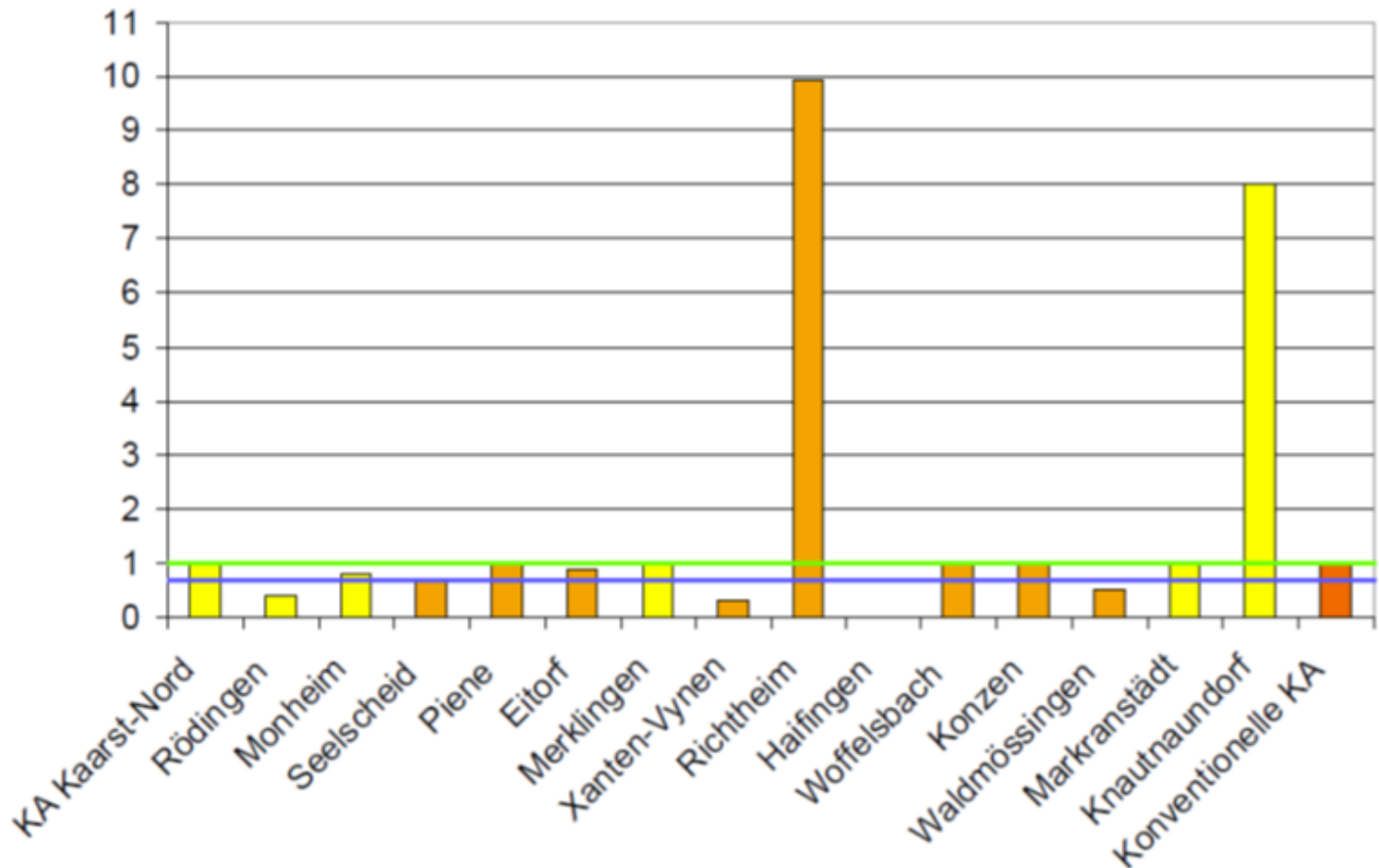
Filtratsammler

Filtrat

Membrant

Aufsaugkanal

B



In practice < 0,3 mg TP/l possible

PINNEKAMP ET AL., (2008)

Thank you for your attention

Contact

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